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## PENTAFLUOROOXYTELLURIUM SUBSTITUTED FLUOROCARBONS

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Pentafluorooxytellurium substituted fluorocarbons (TeF<sub>5</sub>0R<sub>f</sub>), a previously unknown class of compounds, have been prepared by the reactions of either TeF<sub>5</sub>0Cl or TeF<sub>5</sub>0F with fluoroolefins. The addition products are low volatility colorless fluids. Surprisingly, the hypofluorite reacts more controllably and in higher yield (70-85%) than the hypochlorite (20-30%), wherein by-product forming interactions predominate. Addition of TeF<sub>5</sub>0F to the double bonds of unsymmetrical olefins results in isomeric products. Details of the syntheses and the characterization of these compounds will be presented. A comparison of the properties of the perfluorocyclopentene adducts  $XF_50C_5F_6$  will be made for the series X = S, Se, Te.

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## SELENIUM-NITROGEN AND TELLURIUM-NITROGEN COMPOUNDS

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Te-N and Se-N compounds are notoriously unstable. However, starting with  $\rm H_2N$ -TeF $_5$  or  $\rm (CH_3)_3Si-NH-TeF_5$  a variety of tellurium nitrogen compounds have been prepared, such as  $\rm F_5TeN=SF_2$ ,  $\rm F_5Te-N=PF_3$ ,  $\rm F_5Te-NCl_2$  a.o. A crystal structure is given of  $\rm F_5Te-N=WCl_4$ , and an electron diffraction structure of  $\rm F_5Te-N=C=0$ . Whereas  $\rm F_5S-N=C=0$  has a similar structure,  $\rm F_5Se-O-C=N$  appears as cyanate! The first selenium-nitrogen double bond systems have been prepared with  $\rm F_5Te-N=SeCl_2$  and  $\rm F_5Te-N=SeF_2$ .